

## 5-Membered Heterocyclic Allylsilanes in Synthesis : Generation *via* Aza-ene Reaction and Application to the Synthesis of a Bicyclic 1,2-Dinitrogen Analogue of Naturally Occurring Pyrrolizidines

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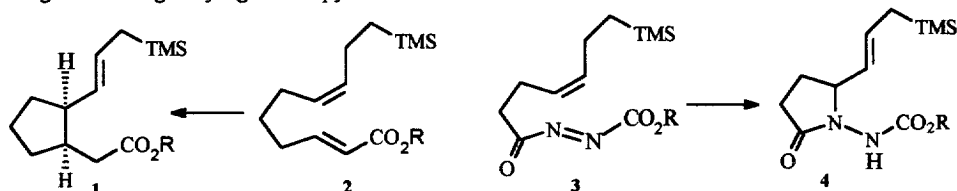
Dedicated to Prof. S. K. Talapatra on the occasion of his 65<sup>th</sup> Birthday

**Abstract:** A cyclic hydrazide containing an allylsilane functionality obtainable by an aza-ene reaction provides ready access to a bicyclic 1,2-dinitrogen compound related to naturally occurring pyrrolizidines. © 1998 Published by Elsevier Science Ltd. All rights reserved.

**Keywords:** Aza-ene Reaction, Heterocyclic Allylsilanes, Fused Tetrahydropyrazole

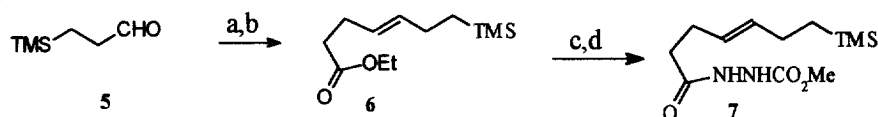
In a series of papers, we have recently described the utility of cyclopentanoid allylsilanes **1** as versatile building blocks for the synthesis of a diverse group of cyclopentane containing natural products (Scheme 1).<sup>1,2</sup> These intermediates are available in near quantitative yields by 5-(3,4) ene cyclization of activated 1,6-dienes **2** containing a homoallylsilane unit as ene donor. This work prompted us to examine a related aza-ene reaction<sup>3</sup> of **3** featuring a reactive azodicarbonyl moiety as enophile to the cyclic hydrazide **4**. The resulting species can then be elaborated *via* reductive cleavage of the N-N bond<sup>3,4</sup> and exploitation of the allylsilane side-chain to azabicyclo[3.3.0]octane, the nucleus of a group of pyrrolizidine alkaloids which continues to receive intense attention by synthetic chemists.<sup>5</sup> In this letter, we report on the feasibility of the aza-ene route to **4** and allylsilane induced stereoselective cyclization of the corresponding methoxymethyl derivative **9** to a fused tetrahydropyrazole **10** which may be looked upon as a mononitrogen analogue of biologically significant pyrrolizidines.

Scheme 1



The ene educt **7** (mp. 76° C), a crystalline acylhydrazocarboxylate, was made from 3-(trimethylsilyl)propanal (**5**)<sup>6</sup> in an overall yield of 51% as described in Scheme 2.

Scheme 2

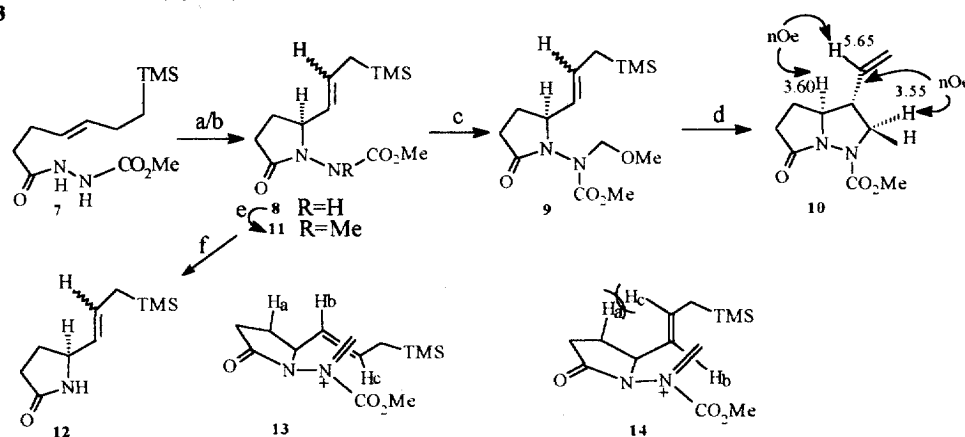


a) CH<sub>2</sub>=CHMgBr, THF, 80%. b) CH<sub>3</sub>C(OEt)<sub>3</sub>, H<sup>+</sup>, tol, 82%. c) KOH, MeOH, 94%. d) (i) NaH, bz, then (COCl)<sub>2</sub>; (ii) H<sub>2</sub>NNHCO<sub>2</sub>Me, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 84%.

It was envisaged that oxidants<sup>7</sup> capable of converting hydrazides into azo compounds would convert **7** into **8** (Scheme 3) *via* ene reaction of the *in situ* generated azodicarbonyl intermediate (*cf.* **3**). It was also deemed necessary to make use of neutral or slightly basic oxidants to ensure survival of the acid labile allylsilane **8**. Two different oxidants *e.g.*, Ag<sub>2</sub>CO<sub>3</sub>-impregnated celite<sup>8</sup> and activated MnO<sub>2</sub><sup>9</sup> were initially selected for optimum results. Sonicating **7** with 25–30 eq of activated MnO<sub>2</sub><sup>3,9</sup> in 1,2-dichloroethane at 15°C for 3h gave a white crystalline solid which upon recrystallization from ether-petroleum ether (40–60°C) gave *E*-**8** (m.p. 89°C) containing traces (*ca.* 5%) of the *Z*-isomer

in an yield of 49%. The structural and stereochemical confirmation of **8**<sup>13</sup> followed from analysis of its <sup>1</sup>H- and <sup>13</sup>C-NMR spectra. Sonication of **7** with 10 eq of Ag<sub>2</sub>CO<sub>3</sub>/celite<sup>3</sup> reagent in hot benzene for 7h followed by preparative layer chromatography of the crude product also gave a semisolid material (38%) as a mixture (the isomers do not resolve on TLC) of *E*-**8** and *Z*-**8** in a ratio of 4.5:1, respectively. It is our experience that MnO<sub>2</sub> oxidations generally give a purer ene product **8** in consistently good yields.

Scheme 3



a) MnO<sub>2</sub>, CICH<sub>2</sub>CH<sub>2</sub>Cl, 15°C, 49%. b) Ag<sub>2</sub>CO<sub>3</sub>-Celite, bz, 38%. c) NaH, MOMCl, THF, 90%. d) BF<sub>3</sub>, OEt<sub>2</sub> (2.5 eq), CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 51%. e) KOBu<sup>t</sup>, MeI, THF, 0°C, 90%. f) Li/NH<sub>3</sub>, 70%.

In order to demonstrate the utility of the allylsilane side-chain, **8** was next converted to the methoxymethyl derivative **9** and exposure of the latter to BF<sub>3</sub>·Et<sub>2</sub>O<sup>10,11</sup> (2.5 eq) in dichloromethane gave the fused tetrahydropyrazole **10** in 51% yield. The structure and relative stereochemistry of **10**<sup>13</sup> rest on high-field <sup>1</sup>H- & <sup>13</sup>C-NMR as well as nOe studies. The high stereoselectivity in the reaction 9→10 is readily explicable in terms of the synclinal<sup>12</sup> transition structure **13** of the N-acylhydrazone intermediate which for steric reasons is largely favoured over **14**. Finally, **8** was converted to **11** and reductive cleavage of the latter with Li/NH<sub>3</sub> yielded **12**, the building block for pyrrolizidine alkaloids.

In conclusion, a route to 5-membered heterocycles **8** and **12**, each with a built-in allylsilane terminator has been developed and the use of the former in the synthesis of a fused tetrahydropyrazole demonstrated. Further work to utilize **12** for the synthesis of azabicyclo[3.3.0]octane skeleta related to pyrrolizidines is under active investigation in this laboratory.

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#### References & Notes

- Sarkar, T. K.; Ghorai, B. K.; Nandy, S. K.; Mukherjee, B.; Bannerji, A. *J. Org. Chem.* **1997**, *62*, 6006 and references cited therein.
- Sarkar, T. K.; Gangopadhyay, P.; Ghorai, B. K.; Nandy, S. K.; Fang, J.-M. *Tetrahedron Lett.* (in press).
- Vedejs, E.; Meier, G. P. *Tetrahedron Lett.* **1979**, *20*, 4185.
- Scartozzi, M.; Grondin, R.; Leblanc, Y. *Tetrahedron Lett.* **1992**, *33*, 5717.
- For a review see: Casiraghi, G.; Zanardi, F.; Rassa, G.; Pinna, L. *Org. Prep. & Procd. Int.* **1996**, *28*, 641.
- Sarkar, T. K.; Ghosh, S. K.; Subba Rao, P. S. V.; Mamdapur, V. R. *Tetrahedron* **1992**, *48*, 6897.
- Mijs, W. J.; Dejonge, C. R. H. I. "Organic Syntheses by oxidation with Metal Compounds." Plenum Press, New York, 1986.
- Fetizon, M.; Golfier, M.; Milcent, R.; Papadakis, I. *Tetrahedron* **1975**, *31*, 165.
- Attenburrow, J.; Cameron, A. F. B.; Chapman, J. H. Evans, R. M.; Hems, B. A.; Jansen, A. B. A.; Walker, T. *J. Chem. Soc.* **1952**, 1094.
- Rutjes, F. P. J. T.; Hiemstra, H.; Mooiweer, H. H.; Speckamp, W. N. *Tetrahedron Lett.* **1988**, *29*, 6975.
- Sarkar, T. K.; Ghorai, B. K.; Das, S. K.; Gangopadhyay, P.; Subba Rao, P. S. V. *Tetrahedron Lett.* **1996**, *37*, 6607.
- Fleming, I.; Barbero, A.; Walter, D. *Chem. Rev.* **1997**, *97*, 2063.
- Data for **8**:  $\nu_{\max}$  1720, 1695, 1650 cm<sup>-1</sup>. <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$ : 7.15(bs, 1H), 5.65(dt, 1H, *J*=15.0 & 8.08 Hz), 5.10(dd, 1H, *J*=15.0 & 8.85 Hz), 4.15 (m, 1H), 3.67(s, 3H), 2.35(m, 2H), 2.22(m, 1H), 1.73(m, 1H), 1.45(dt, 2H, *J*=8.24 & 1.37 Hz), -0.05(s, 9H). <sup>13</sup>C NMR (75MHz, CDCl<sub>3</sub>)  $\delta$ : 173.74(s), 155.65(s), 133.22(d), 126.34(d), 62.05(q), 52.67(d), 28.02(t), 24.74(t), 22.82(t), -2.19(q, 3C). <sup>13</sup>C NMR for *Z*-isomer (partial): 131.87(d), 125.51(d), 55.61(d), 24.45(t), 18.79(t). Anal. Calcd. for C<sub>12</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>Si: C 53.33, H 8.14, N 10.37. Found: C 55.36, H 8.08, N 10.45. Data for **10**:  $\nu_{\max}$  (GC-FTIR at 250 °C) 1772, 1737 cm<sup>-1</sup>. <sup>1</sup>H NMR (360MHz, CDCl<sub>3</sub>)  $\delta$ : 5.65(m, 1H), 5.25(m, 2H), 3.80(m, 1H), 3.80(s, 3H), 3.60(m, 1H), 3.55(m, 1H), 2.67(m, 1H), 2.55(m, 1H), 2.45(m, 1H), 2.38(m, 1H), 1.93(m, 1H). <sup>13</sup>C NMR (90MHz, CDCl<sub>3</sub>)  $\delta$ : 178.8(s), 157.40(s), 133.30(d), 119.20(t), 62.80(d), 53.90(t), 53.70(q), 50.30(d), 27.90(t), 18.90(t). MS *m/z* 210(M<sup>+</sup>, 30%), 127(100%).